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10/551,654	07/10/2006	Tetsuya Okano	0425-1218PUS1	5662	
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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail $\,$ address(es):

mailroom@bskb.com

Application No. Applicant(s) 10/551.654 OKANO ET AL. Office Action Summary Examiner Art Unit ABIGAIL FISHER 1616 -- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --Period for Reply A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS. WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION. Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication. If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication - Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b). Status 1) Responsive to communication(s) filed on 17 November 2008. 2a) This action is FINAL. 2b) This action is non-final. 3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under Ex parte Quayle, 1935 C.D. 11, 453 O.G. 213. Disposition of Claims 4) Claim(s) 4.6-8.10.12-14 and 16-19 is/are pending in the application. 4a) Of the above claim(s) is/are withdrawn from consideration. 5) Claim(s) _____ is/are allowed. 6) Claim(s) 4, 6-8, 10, 12-14, 16-19 is/are rejected. 7) Claim(s) _____ is/are objected to. 8) Claim(s) _____ are subject to restriction and/or election requirement. Application Papers 9) The specification is objected to by the Examiner. 10) The drawing(s) filed on is/are; a) accepted or b) objected to by the Examiner. Applicant may not request that any objection to the drawing(s) be held in abevance. See 37 CFR 1.85(a). Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d). 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152. Priority under 35 U.S.C. § 119 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received.

1) Notice of References Cited (PTO-892)

Notice of Draftsperson's Patent Drawing Review (PTO-948)

information Disclosure Statement(s) (PTO/S5/06)
Paper No(s)/Mail Date ______.

Attachment(s)

Interview Summary (PTO-413)
Paper No(s)/Mail Date.

6) Other:

5) Notice of Informal Patent Application

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DETAILED ACTION

Receipt of Amendments/Remarks filed on November 17 2008 is acknowledged. Claims 1-3, 5, 9, 11, 15 and 20-21 were/stand cancelled. Claims 4, 7-8, 10, 13-14 and 17-18 were amended. Claims 4, 6-8, 10, 12-14, 16-19 are pending.

Rejections and/or objections not reiterated from previous office actions are hereby withdrawn. The following rejections and/or objections are either reiterated or newly applied. They constitute the complete set presently being applied to the instant application.

Claim Rejections - 35 USC § 101

35 U.S.C. 101 reads as follows:

Whoever invents or discovers any new and useful process, machine, manufacture, or composition of matter, or any new and useful improvement thereof, may obtain a patent therefor, subject to the conditions and requirements of this title.

The rejection of claims 20-21 under 35 U.S.C. 101 because the claimed recitation of a use, without setting forth any steps involved in the process, results in an improper definition of a process, i.e., results in a claim which is not a proper process claim under 35 U.S.C. 101 is **withdrawn** in light of Applicants' cancellation of the claim in the reply filed on November 17 2008.

Claim Rejections - 35 USC § 112

The following is a quotation of the second paragraph of 35 U.S.C. 112:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

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The rejection of claims 11 and 20-21 under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention is **withdrawn** in light of applicants' cancellation of the claims in the reply filed on November 17 2008.

Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

The rejection of claims 1-9 and 13-21 under 35 U.S.C. 102(b) as being anticipated by Montgomery (US Patent No. 6221341, cited on PTO Form 1449) is withdrawn in light of Applicants' amendment filed on November 17 2008 amending the pH to be in the range of 1 to 5.

The rejection of claims 1-9, 13-18 and 20-21 under 35 U.S.C. 102(b) as being anticipated by Speed et al. (US Patent No. 6399564) is **withdrawn** in light of Applicants' amendment filed on November 17 2008 amending the pH to be in the range of 1 to 5.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and

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the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

- Applicant Claims
- Determining the scope and contents of the prior art.
- Ascertaining the differences between the prior art and the claims at issue, and resolving the level of ordinary skill in the pertinent art.
- 4. Considering objective evidence present in the application indicating obviousness or nonobviousness

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

The rejection of claims 10-12 under 35 U.S.C. 103(a) as being unpatentable over Montgomery in view of Karlson (US Patent No. 4517159) is withdrawn in light of Applicants' amendment filed on November 17 2008 amending the pH to be in the range of 1 to 5.

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The rejection of claims 10-12 under 35 U.S.C. 103(a) as being unpatentable over Speed et al. in view of Karlson is **withdrawn** in light of Applicants' amendment filed on November 17 2008 amending the pH to be in the range of 1 to 5.

New Rejections Necessitated by the Amendments filed November 17 2008

Claims 4, 6-8, 10, 12-14 and 16-19 are rejected under 35 U.S.C. 103(a) as being unpatentable over Tamura et al. (US Patent No. 5827447) in view of Kobayashi et al. (US Patent No. 5869440).

Applicant Claims

The instant application claims a sterilizer composition having a pH value of 1 to 5 at 25 °C and comprising water and an organic peracid obtained by reacting (A) an ester of a polyhydric alcohol and an organic acid having a hydrocarbon group which may have a hydroxyl group with (B2) hydrogen peroxide in an (A)/(B1) molar ratio of 1/10 to 20/1 in water at pH 8 to 12. The instant application claims a method of sterilizing a material comprising contacting the material with the solution above. The instant application claims a process for producing a sterilizer composition comprising reacting (A) with (B1) in an (A)/(B1) molar ratio of 1/10 to 20/1 in water at pH 8 to 12 then adjusting the reaction system to pH 1 to 5.

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Determination of the Scope and Content of the Prior Art (MPEP §2141.01)

Tamura et al. is directed to a transparent liquid bleaching agent having a transparent appearance and excellent storage stability and bleaching power (abstract). The composition comprises hydrogen peroxide, a surfactant, and a bleach activator capable of yielding an organic peracid with reacted with hydrogen peroxide (column 3, lines 11-15). The hydrogen peroxide is incorporated in an amount of 0.3 to 30% by weight into the composition (column 3 lines 44-46). The bleach activator utilized to yield and organic peroxide when reacted with hydrogen peroxide includes triacetin, a fatty acid anhydride having 2 to 18 carbon atoms, and sodium alkanoyloxybenzenesulfonate (column 6, lines 50-57). Surfactant and bleach activator are incorporated in total amount of (surfactant plus bleach activator) 0.1 to 50% by weight of the composition. The ratio of surfactant to bleach activator is 50:1 to 11 (column 8, lines 60-67). Exemplified ratios of bleach activator to hydrogen peroxide are 2:5 and 1:5 (table 1 and 2). Exemplified methods of formation teach adding surfactants, bleach activators, hydrogen peroxide. The pH value of each composition was adjusted to 2 with sulfuric acid. Commercially available products which showed little storage stability had a pH of 10.2 (column 13, lines 1-2 and table 1). The exemplified compositions all comprise water.

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Ascertainment of the Difference Between Scope the Prior Art and the Claims (MPEP §2141.012)

Tamura et al. do not specify a pH of 8 to 12. Tamura et al. do not specify maintaining at a pH of 8 to 12 for 1 to 120 minutes. Tamura et al. do not teach a method of sterilizing. However, these deficiencies are cured by Kobayashi et al.

Kobayashi et al. is directed to peroxide activation. It is taught that for domestic or industrial use, it is preferable to use hydrogen peroxide of concentrations not higher than 6% in order not to be designated as a powerful drug. It is taught that to stabilize a hydrogen peroxide solution, the pH of the solution is adjusted not higher than 9 (column 2, lines 25-38). It is taught that bleach concentrations lower than 0.3 wt. %, the bleaching effect is not sufficient (column 3, lines 26-27). It is taught that an alkaline pH is utilized to make hydrogen peroxide decompose so as to be used as an agent for cleaning, bleaching, sterilizing, deodorizing etc. in domestic or industrial use (column 3, lines 17-21). As shown in comparative example 4, when utilizing triacetin as the bleach activator, when the pH is maintained at 10.9 there is a 80% reduction of the bleaching rate at 50 °C (table 2, example 4).

Finding of Prima Facie Obviousness Rationale and Motivation (MPEP §2142-2143)

It would have been obvious to one of ordinary skill in the art at the time of the instant invention to combine the teachings of Tamura et al. and Kobayashi et al. and utilize a pH greater than 9 in order to decompose the hydrogen peroxide for use as a sterilizer but then decrease the pH in order to stabilize the hydrogen peroxide solution

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for long term storage. One of ordinary skill in the art would have been motivated to initially have the pH of a peracid generating solution be greater than 8 as Kobayashi et al. teach that this is the pH utilized to decompose the hydrogen peroxide so it can be utilized in domestic or industrial use. Then one of ordinary skill in the art would have been motivated to decrease the pH to that of around 2 as Tamura et al. teach that this is a pH that provides long-term shelf stability and Kobayashi et al. teach that triacetin and hydrogen peroxide containing solutions lose about 80% of their efficacy after 5 days at a pH of around 11. While Tamura et al. do not specify the pH of the composition initially, it must necessarily be higher than 2 because the pH is adjusted to 2 with sulfuric acid. Therefore, based on these teachings it would have been obvious to one of ordinary skill in the art to initially have the pH of the bleaching composition be greater than 8 and then decrease the pH to that of about 2.

It would have been obvious to one of ordinary skill in the art at the time of the instant invention to combine the teachings of Tamura et al. and Kobayashi et al. and utilize the bleaching composition in a method of sterilization. One of ordinary skill in the art would have been motivated to utilize the bleaching composition in a method of sterilization as this type of method is taught as a suitable use for a bleaching composition by Kobayashi et al.

Regarding the structure of product (A) (specifically corresponding to claims 1, 7-8, 10, 13-14 and 17-18), the instant specification indicates compounds fitting this particular structure include triacetin. Triacetin is a specific type of bleach activator taught by Tamura et al.

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Regarding the claim amount of hydrogen peroxide, Tamura et al. teach amounts that overlap that instantly claimed. In the case where the claimed ranges "overlap or lie inside ranges disclosed by the prior art" a *prima facie* case of obviousness exists. **See**MPEP 2144.05 [R-5]

Regarding the claimed reaction time of the hydrogen peroxide and triacetin at pH greater than 8, it would have been obvious to one of ordinary skill in the art to manipulate the reaction time in order to optimize the decomposition product while maintaining bleaching efficacy. Kobayashi et al. teach that a triacetin and hydrogen peroxide solution losses about 80% efficacy after 5 days and teaches that some decomposition of the hydrogen peroxide is necessary in order for the agent to be utilized. Therefore, it would have been obvious to one of ordinary skill in the art to optimize the reaction time in order to what time produces the optimal bleaching effect. It would have been obvious to one of ordinary skill in the art at the time of the invention to engage in routine experimentation to determine optimal or workable ranges that produce expected results. Where the general conditions of a claim are disclosed in the prior art, it is not inventive to discover the optimum or workable ranges by routine experimentation. In re Aller, 220 F. 2d 454, 105 USPQ 233 (CCPA 1955).

Absent any evidence to the contrary, and based upon the teachings of the prior art, there would have been a reasonable expectation of success in practicing the instantly claimed invention. Therefore, the invention as a whole would have been *prima facie* obvious to one of ordinary skill in the art at the time the invention was made.

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Conclusion

No claims are allowed.

Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to ABIGAIL FISHER whose telephone number is (571)270-3502. The examiner can normally be reached on M-Th 9am-6pm EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Johann Richter can be reached on 571-272-0646. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

Abigail Fisher Examiner Art Unit 1616

ΑF

/Mina Haghighatian/ Primary Examiner, Art Unit 1616